

Exploring the reactivity of $n\text{-C}_4\text{-C}_6$ alcohols: from JSR and RCM experiments to operating regimes in a HCCI engine

M. Pelucchi¹, S. Namysl², M. Bissoli¹, K.P. Somers³,

F. Battin-Leclerc², H.J. Curran³, T. Faravelli¹

¹ *Department of Chemistry, Materials and Chemical Engineering, Politecnico di Milano, Italy*

² *CNRS, Université de Lorraine, ENSIC, Nancy Cedex, France*

³ *Combustion Chemistry Center, National University of Ireland, Galway, Ireland*



POLITECNICO
MILANO 1863



OÉ Gaillimh
NUI Galway



1. New Ignition Delay Time data for C₂-C₅ linear alcohols

NUIG Rapid Compression Machine (T=700-925 K, p=10-30 bar)

2. New JSR data for C₄-C₆ linear alcohols

CNRS Jet Stirred Reactor (T=500-1100 K, p=1.06 bar)

3. Update and extension of the POLIMI kinetic model for alcohols LT combustion

4. Evaluation of operability maps in HCCI engine

n-butanol, *n*-pentanol

5. Relevant pathways underlying the auto-ignition phenomena



“This project has received funding from the EU’s Horizon 2020 research and innovation programme under grant agreement N° 723706”.

New Fuel Requirements

- **Compatible** with existing infrastructures (engines, distribution etc.): **ignition**, miscibility, **flame speed**, viscosity, boiling point, corrosive potential etc.
- **Cleaner**
- **Sustainable** production

Fuel	LHV (MJ/L)	RON	MON	CN
Gasoline	~30-33	88-98	80-88	n.d.
Diesel	~35	n.d.	n.d.	40-55
Ethanol	21.4	109	90	n.d.
n-butanol	26.9	98	85	12
n-pentanol	28.5	80	74	20
n-hexanol	29.3	56	46	24

Sarathy et al. (*Prog. Ener. Combust. Sci.*, 2014) and Kalghatgi (*Int. J. Engine Res.*, 2014)

New Fuels

Today (% v/v)

Ethanol: 5-85% in Gasoline

Tested

n-butanol: 13-26% in Gasoline
up to 24% in Diesel

n-pentanol: 10-30% in Diesel

Trends

Increasing Biofuel content

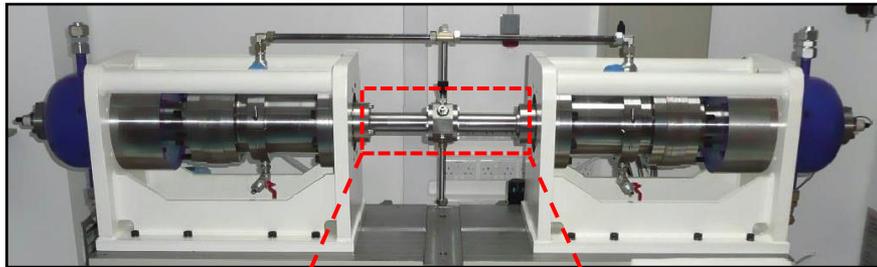
Co-Design of Fuels and Engines

Alcohols show promising physical and chemical properties!

Ignition Delay Time Measurements in RCM: *n*-butanol and *n*-pentanol



Rapid Compression Machine



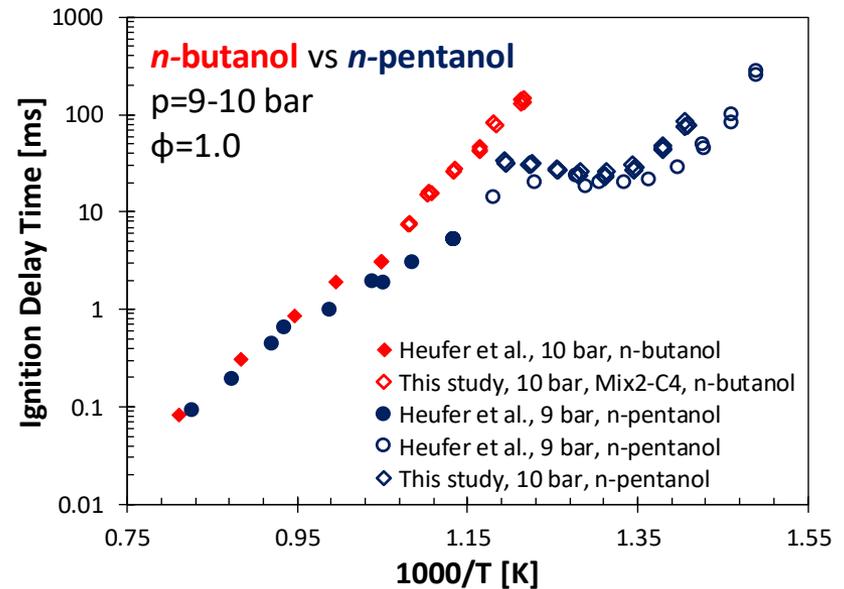
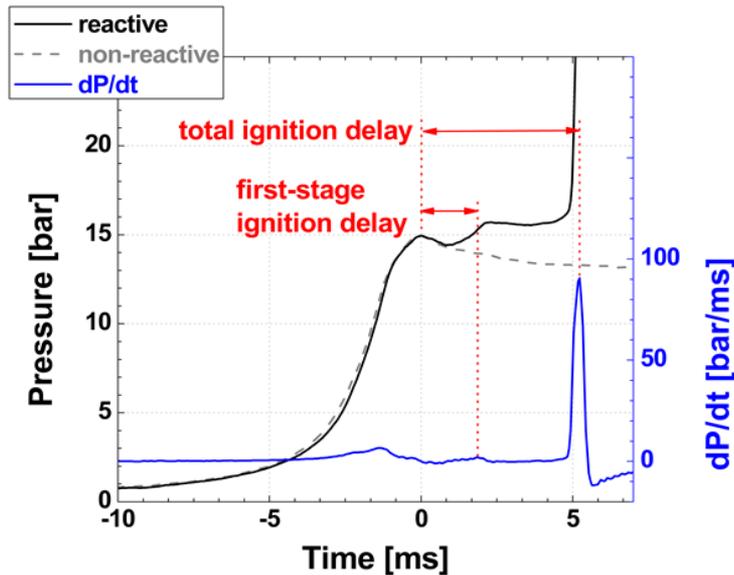
piston
motion



Fuel/air, $\Phi=1.0$, mole %

<i>n</i> -butanol						
Mixture	Fuel	O ₂	N ₂	Ar	CO ₂	P[bar]
1-C ₄	3.38	20.29	68.70	7.63	0.00	30
2-C ₄	3.38	20.28	15.26	61.09	0.00	10
3-C ₄	3.39	20.36	53.61	0.00	22.63	30
<i>n</i> -pentanol						
Mixture	Fuel	O ₂	N ₂	Ar	CO ₂	P[bar]
1-C ₅	2.72	20.43	76.85	0.00	0.00	10
2-C ₄	2.72	20.43	38.42	38.42	0.00	10

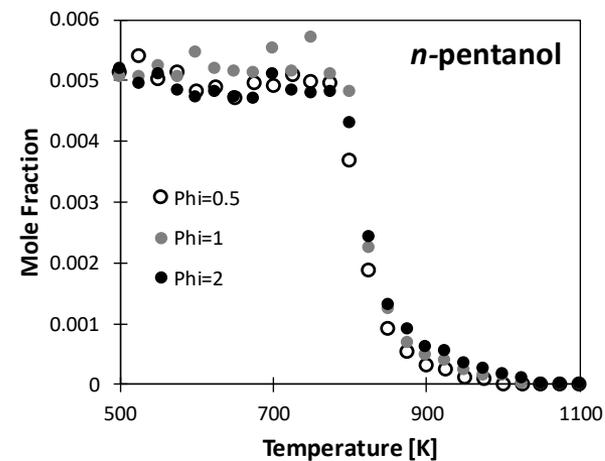
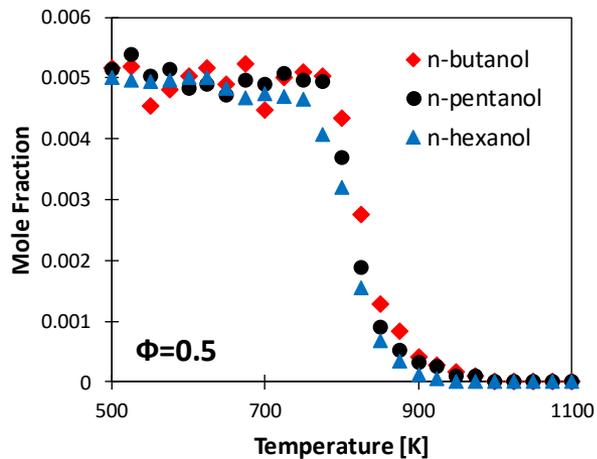
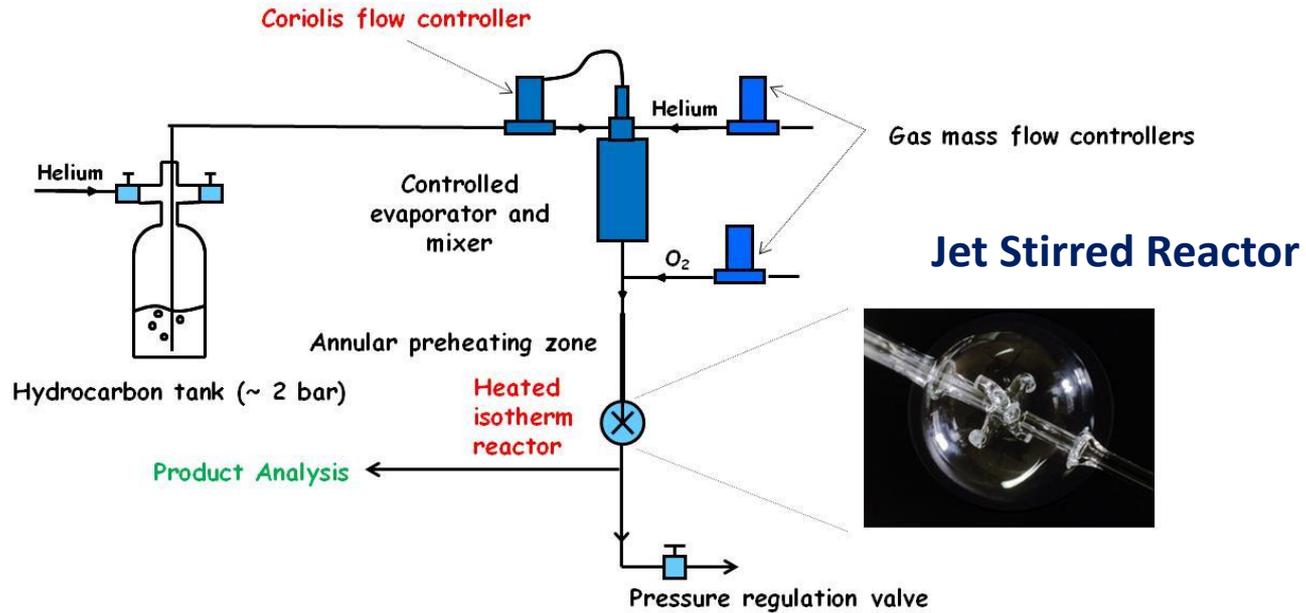
Pelucchi et al. *SAE Int. J. Engine*, 2017



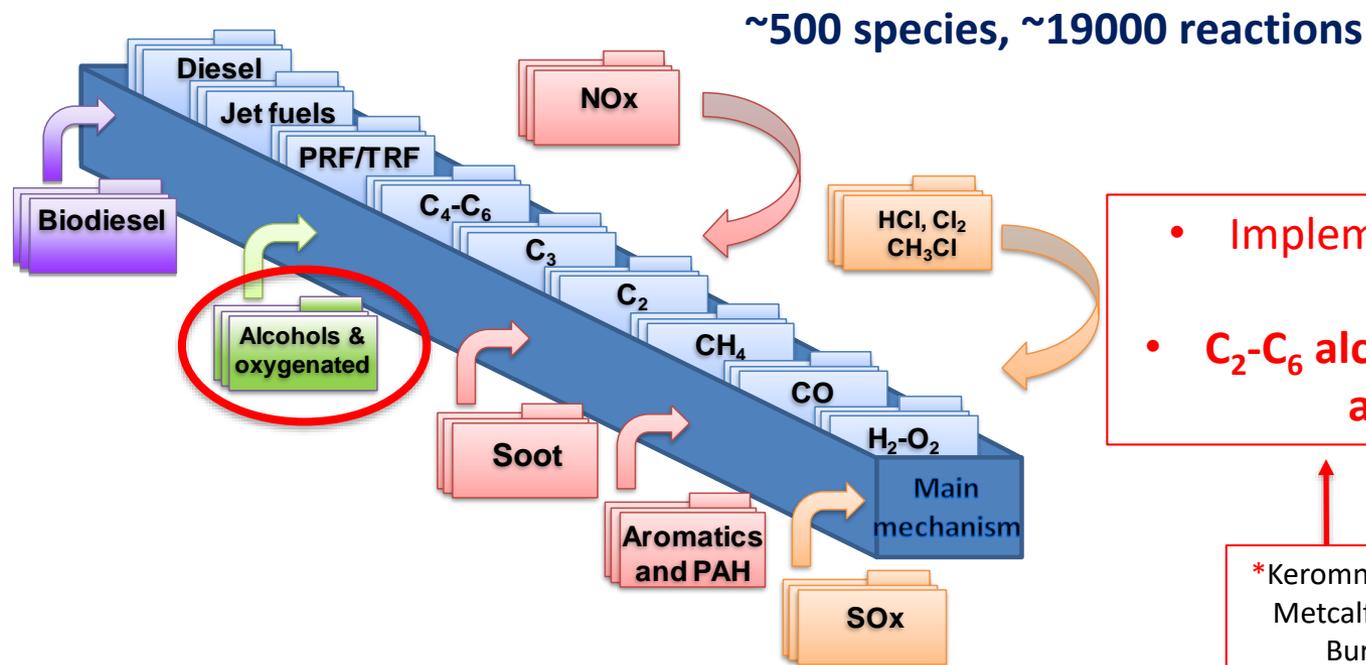
Heufer et al. *Proc Combust Inst*, 2013

Species Measurements in JSR: *n*-butanol and *n*-pentanol

0.5% fuel/O₂/He mixtures (*n*-butanol, *n*-pentanol); $\Phi=0.5, 1.0, 2.0$; $p=1.06$ bar; $\tau=2$ s.



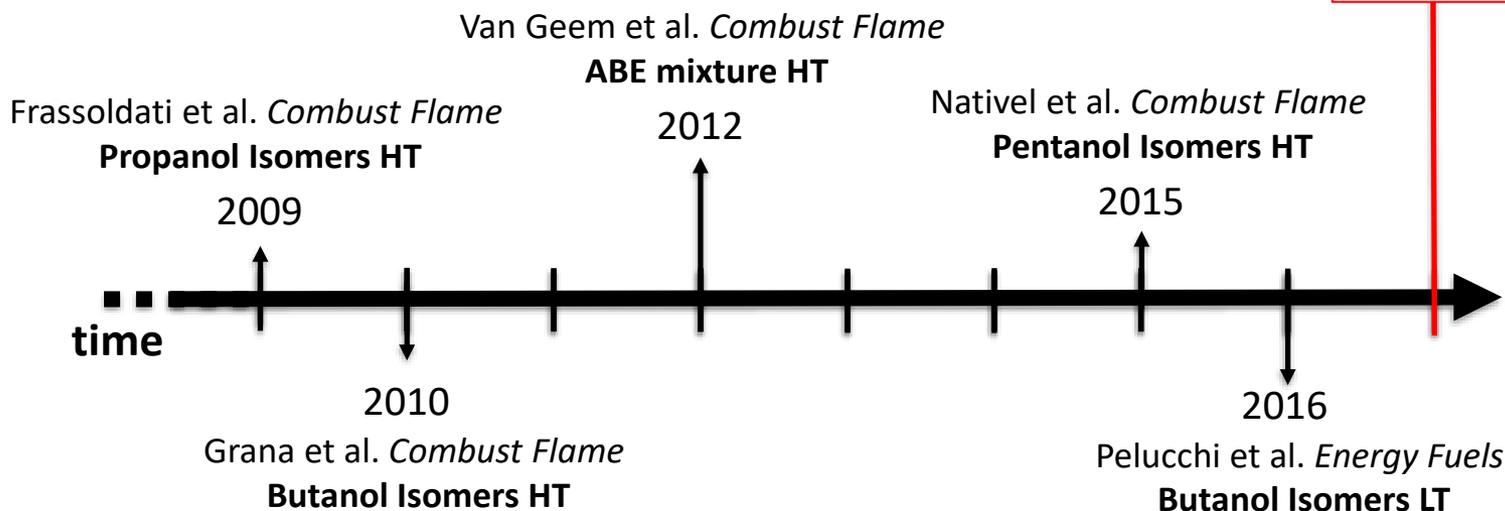
POLIMI Kinetic Mechanism: alcohols module and recent updates



2017

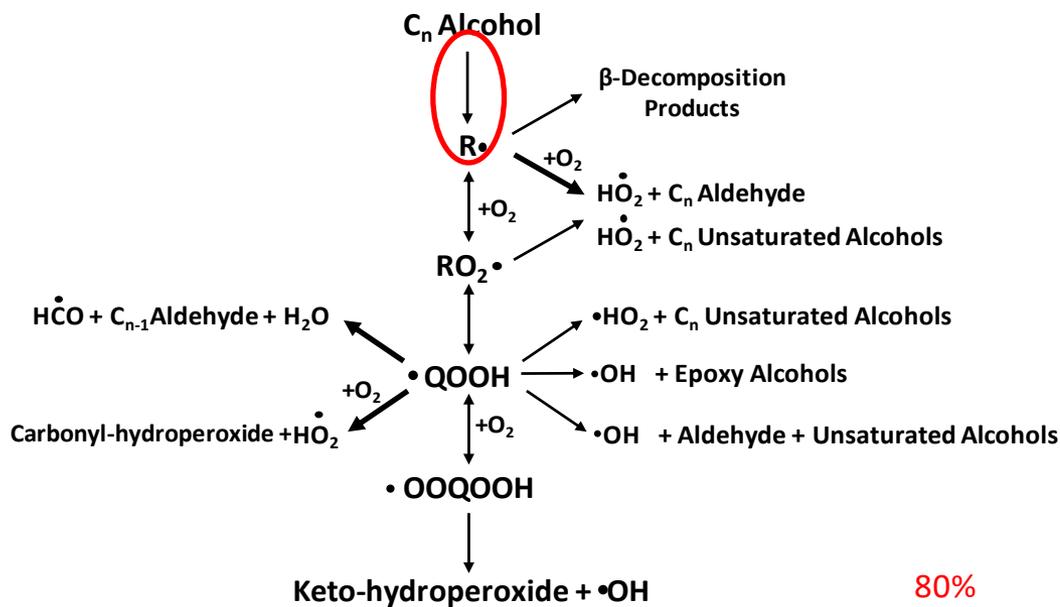
- Implementation of NUIG C_0 - C_2 mech*
- C_2 - C_6 alcohols LT extension and revision

*Keromnes et al. *Combust Flame* 2013
 Metcalfe et al. *Int. J. Chem. Kin.* 2013
 Burke et al. *Combust Flame* 2014

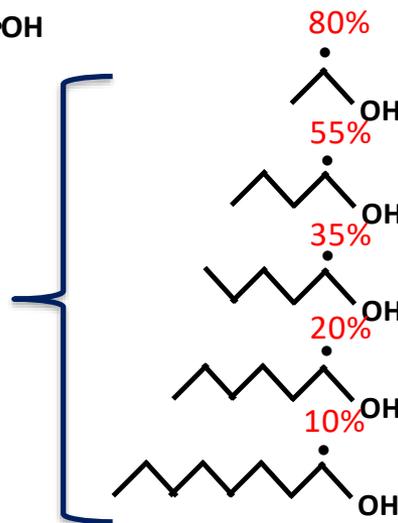


Kinetic model of *n*-butanol and *n*-pentanol LT combustion

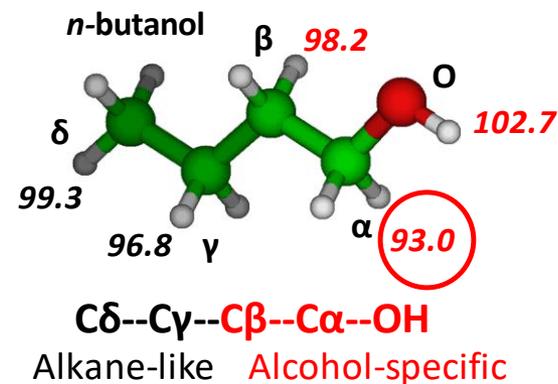
Low Temperature Oxidation Pathways



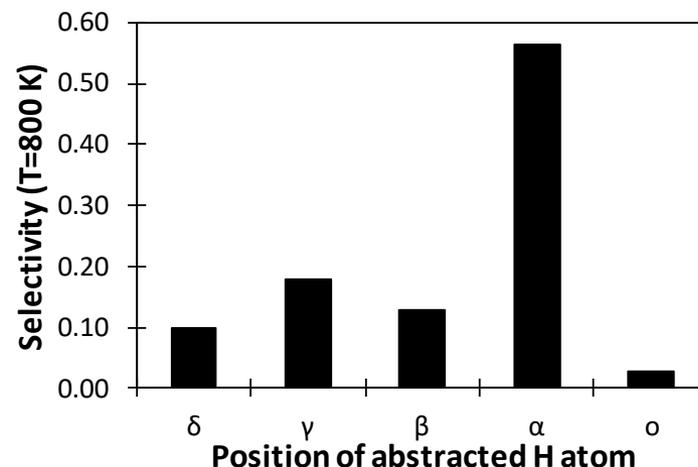
C_n alcohol → *C_n alkane*
for increasing *n*



G4 BDE @ 298 K

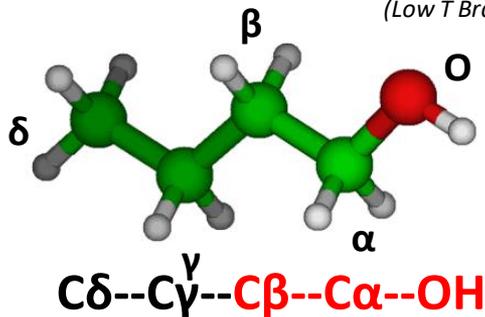
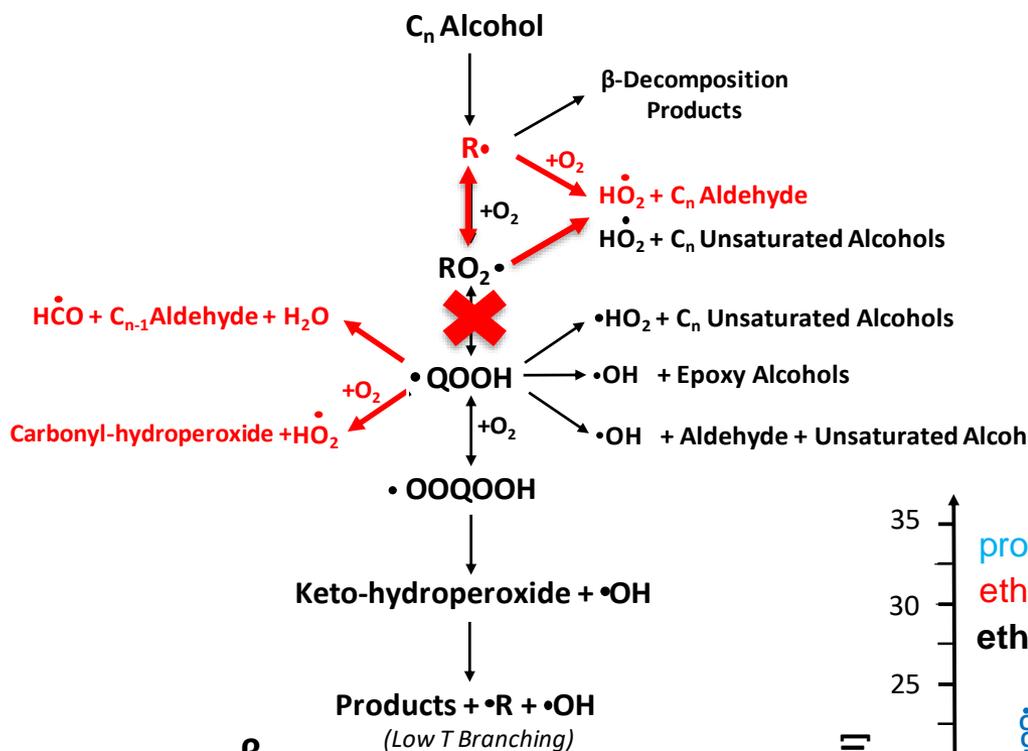


OH + *n*-butanol = H₂O + R

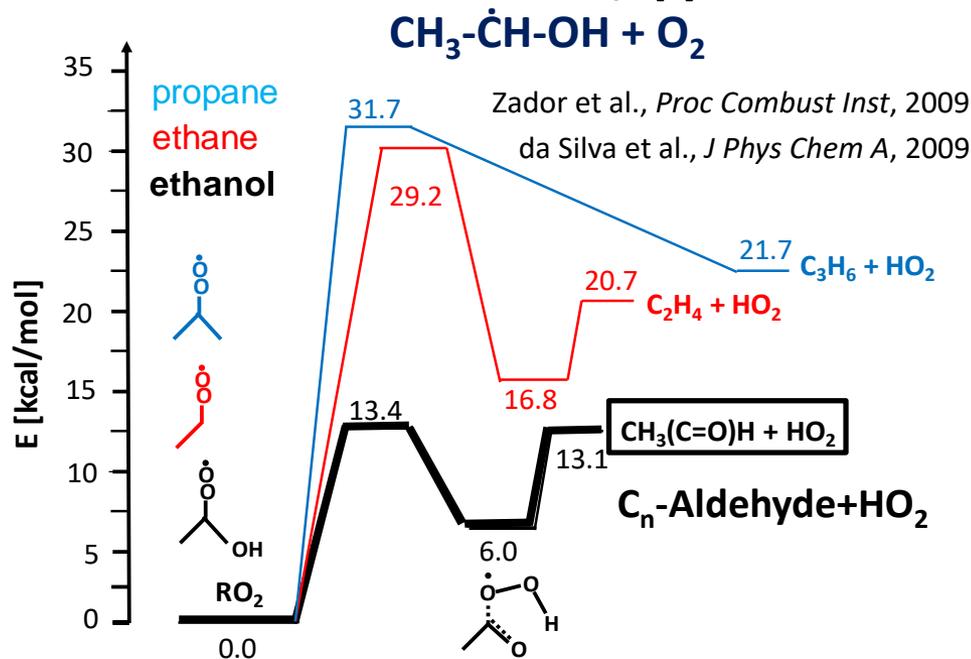
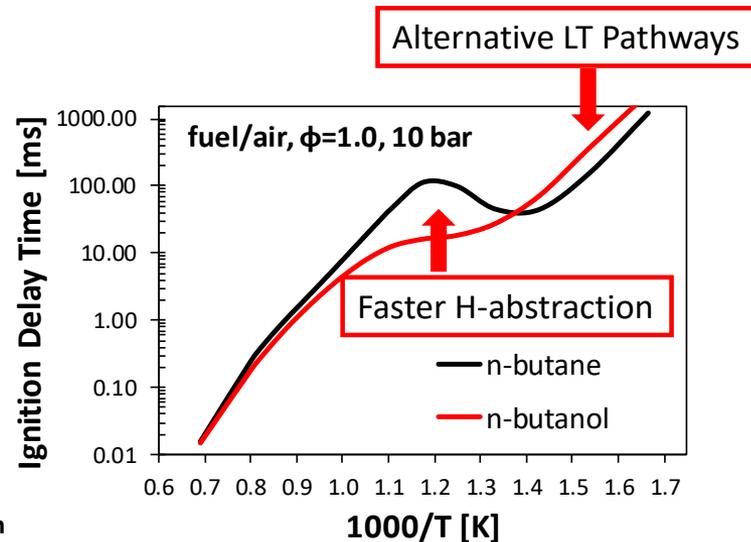


Kinetic model of *n*-butanol and *n*-pentanol LT combustion

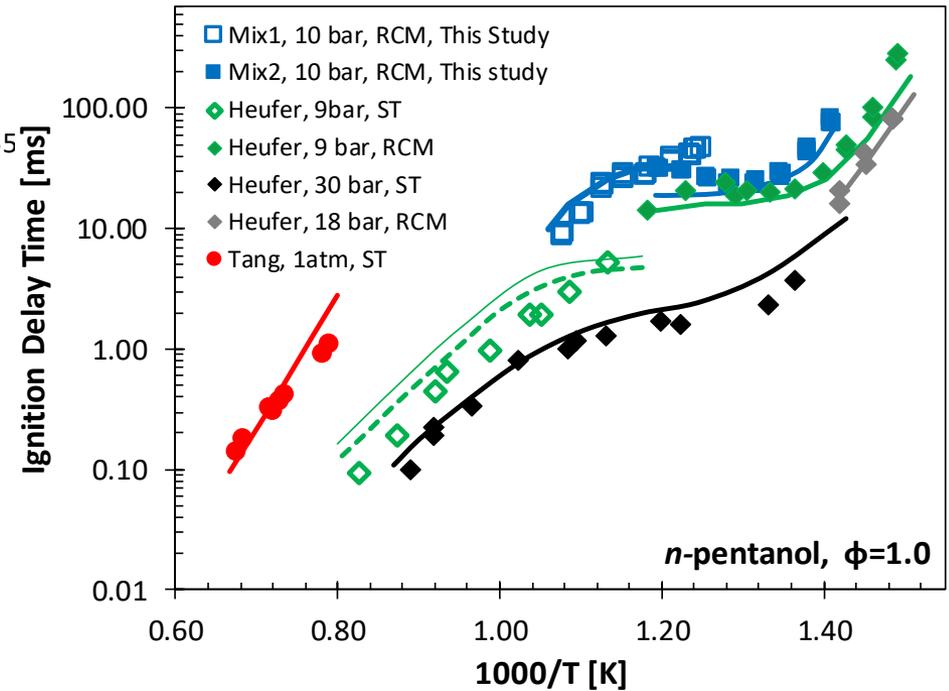
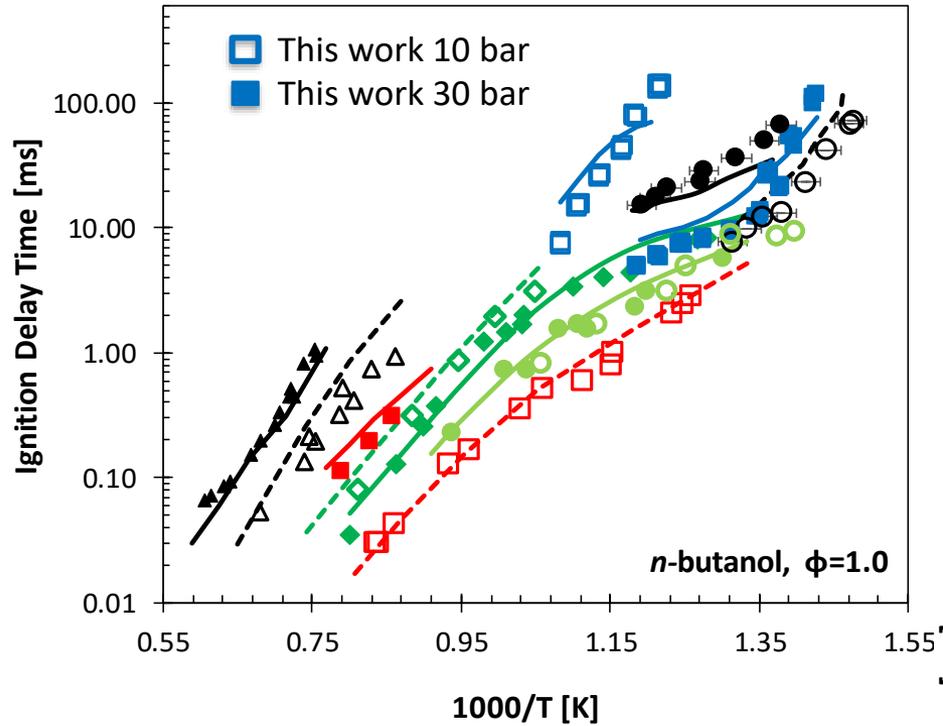
Low Temperature Oxidation Pathways



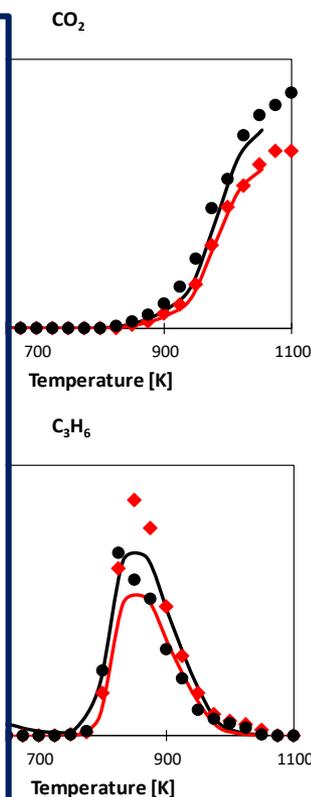
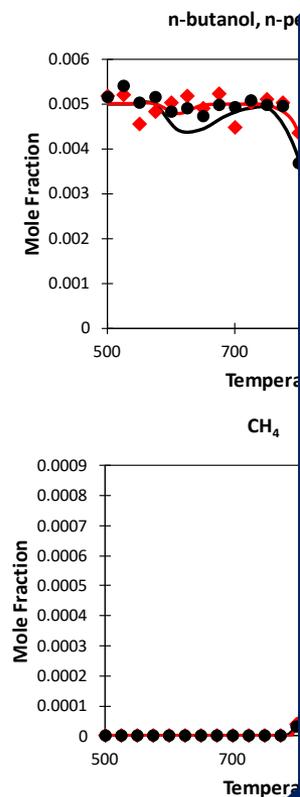
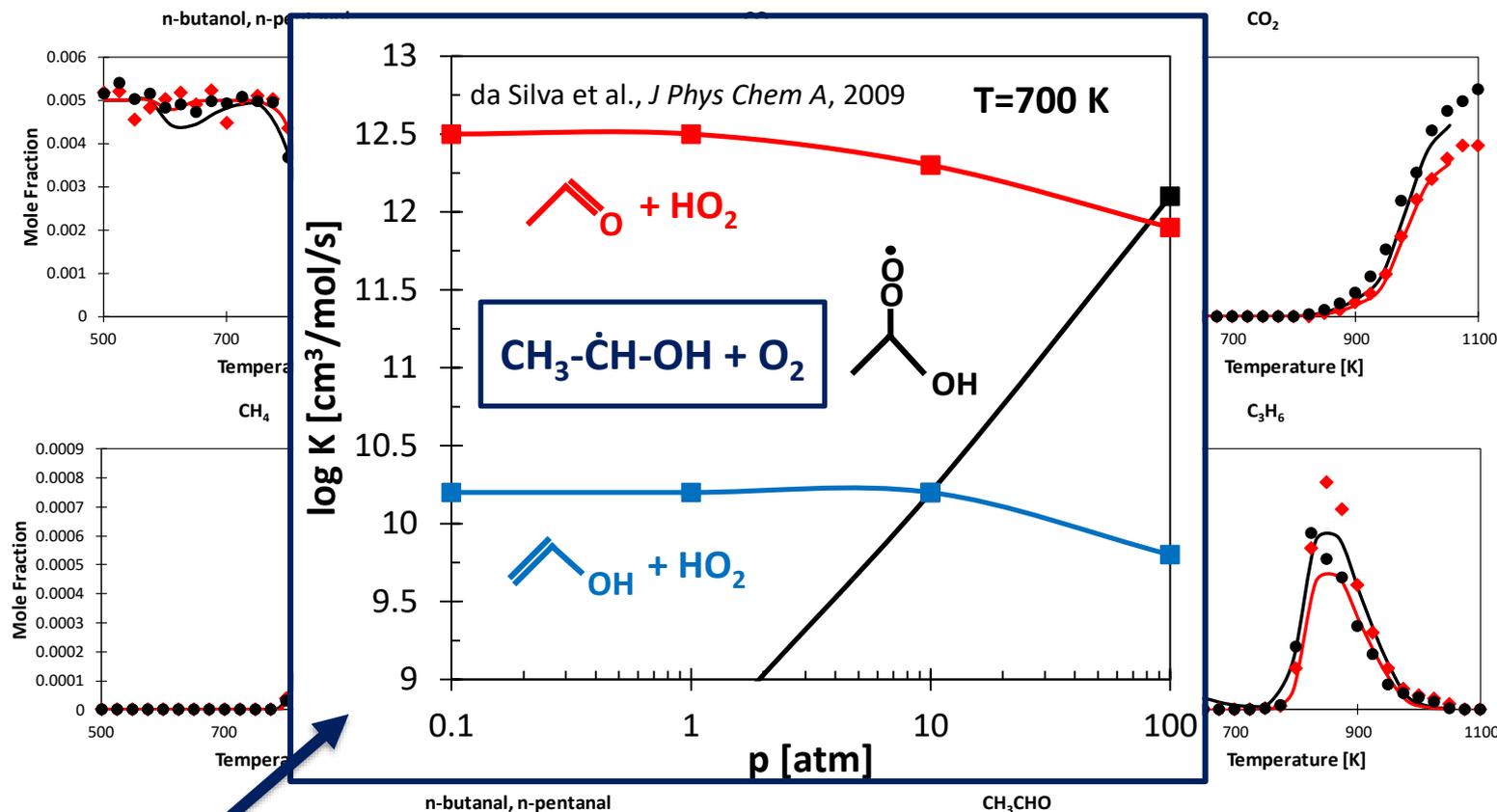
Alkane-like Alcohol-specific



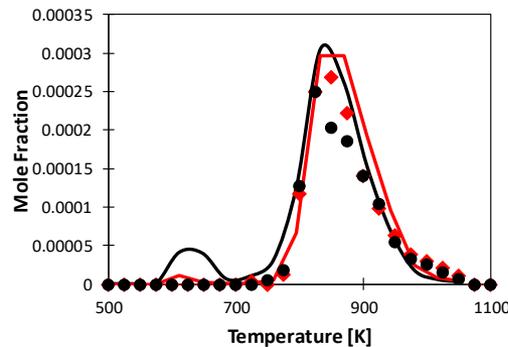
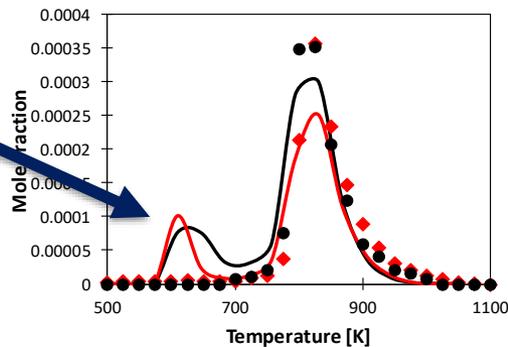
Kinetic model of *n*-butanol and *n*-pentanol: validation, ST / RCM data



Kinetic model of *n*-butanol and *n*-pentanol: validation, JSR data



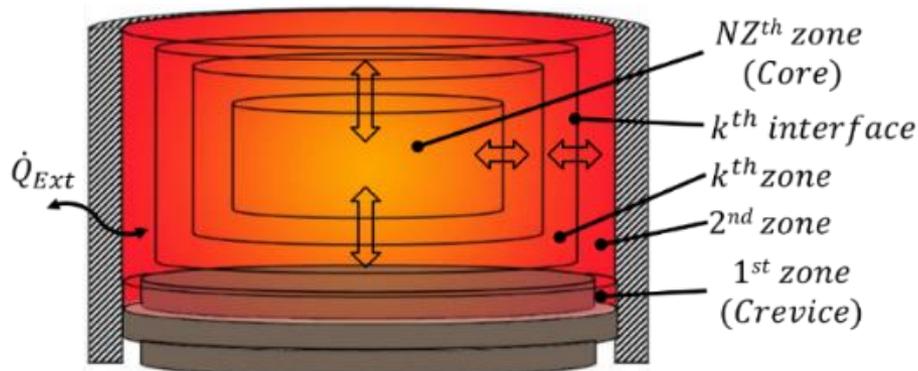
Pressure dependence of $\text{R}\alpha + \text{O}_2$ pathway?



- **HCCI:** Homogeneous Charge Compression Ignition
- **RCCI:** Reactivity Controlled Compression Ignition
- **PCI:** Premixed Compression Ignition
- **GCI:** Gasoline Compression Ignition

POLIMI multi-zone model (ICE)

Bissoli et al. *SAE Int. J. Engines*, 2013; *Appl. Energ.*, 2016; *Energ. Fuel.*, 2017



✓ **Conceived to handle detailed/semi-detailed chemistry**

OpenSMOKE++ framework

Cuoci et al. *Comp. Phys. Comm.*, 2015

- Ideal reactors
- RCM
- Laminar Flames
- Droplets
- Surface Reactor
- ...

Multi-zone model of HCCI Engine: OpenSMOKE++ ICE Model

✓ General structure

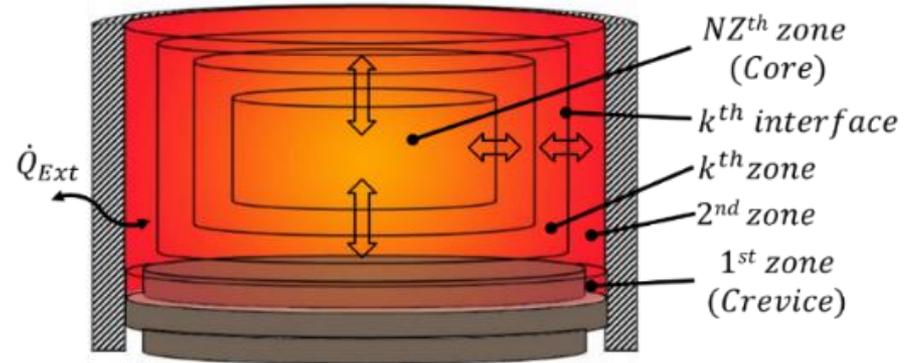
- Multi-Zone “onion-like” structure
- Mixture treated as an **ideal gas**
- **Velocity field not solved**
- **From IVC to EVO**

✓ Zone structure

- Well-mixed reactors, $V(t)$, uniform P
- **Transport among zones** (laminar and turbulent)
- **Crevice**: fixed-volume variable-mass zone

✓ Sub-models

- **Wall heat transfer**
 - **Turbulence**
 - Residual Burned Gases (RBG): cycle-to-cycle predictions
 - **Tools for kinetic analyses**
- } Non-reactive CFD used to define **3 engine-specific parameters (C_{u_z} , C_x , C_{u_w})**



Validation and further model details

Bissoli et al. *SAE Int. J. Engines*, 2013; *Appl. Energ.*, 2016; *Energy Fuels*, 2017

Multi-zone model of HCCI Engine: Operability Maps

2-D operative maps identify stable operability regions

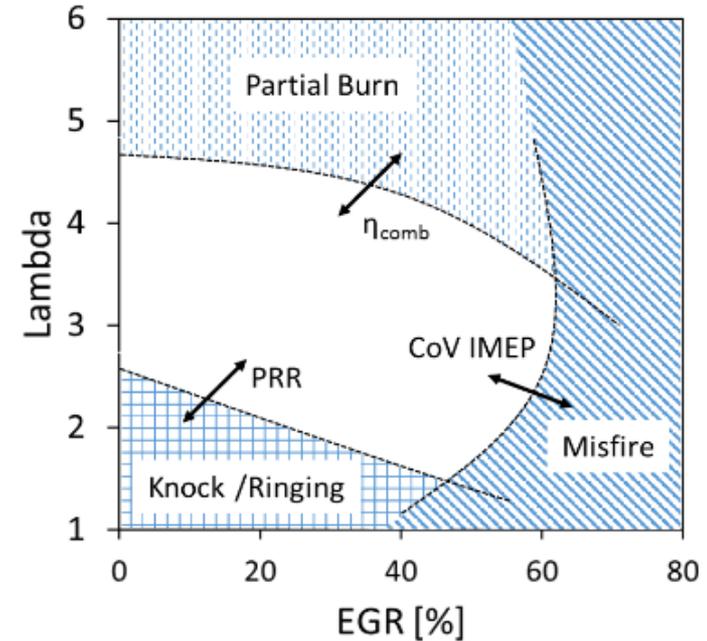
i.e. set of different λ -EGR conditions satisfying the constraints below:

Limits

- Partial Burn: $\eta_{\text{COMB}} \geq 80\%$
- Knock/Ringing: $\text{PRR} \leq 6$ [bar/deg]
- Misfire: $\text{CoV IMEP} \leq 60\%$

$$\eta_{\text{comb}} = \frac{\int HRR}{\Delta H_{\text{comb}}}$$

$$\text{IMEP} = \frac{\int p dV}{V_{\text{disp}}} \quad \text{CoV IMEP} = \frac{\sigma_{\text{IMEP}}}{\text{IMEP}}$$



Ricardo E6 engine Brunel University^[2]

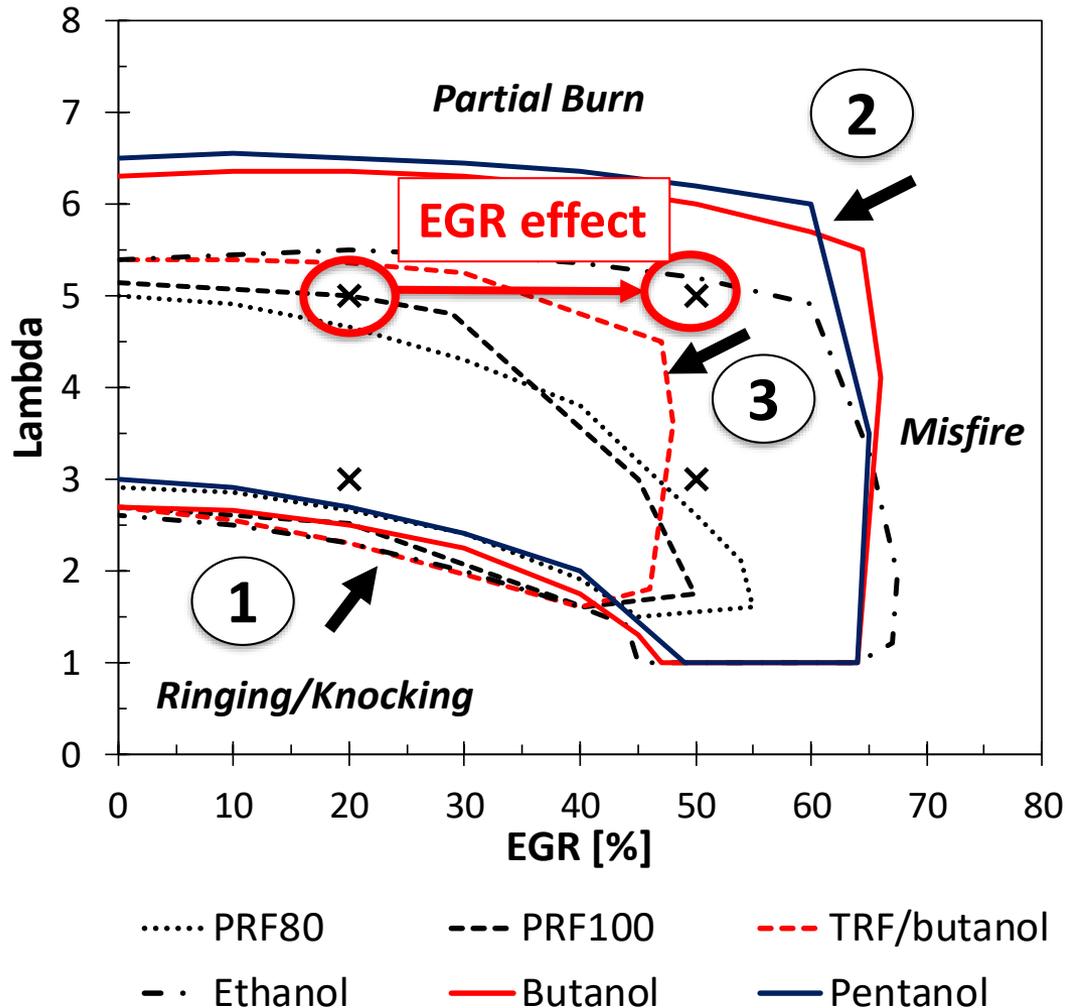
Displacement	504 cm³	CR	11.5
Bore	76.2 mm	Speed	1500 rpm
Stroke	111.1 mm	IVC	137 °ATDC
Rod Length	241.3 mm	EVO	144 °BTDC

Model configuration

15 zones (14+crevice)
 50 cycles (EGR effects)
 >150 λ -EGR combinations
 for each map

[1] Bissoli et al. *Energy Fuels*, 2017; [2] Oakley, PhD Thesis, 2001

Multi-zone model of HCCI Engine: Operability Maps



General Trends

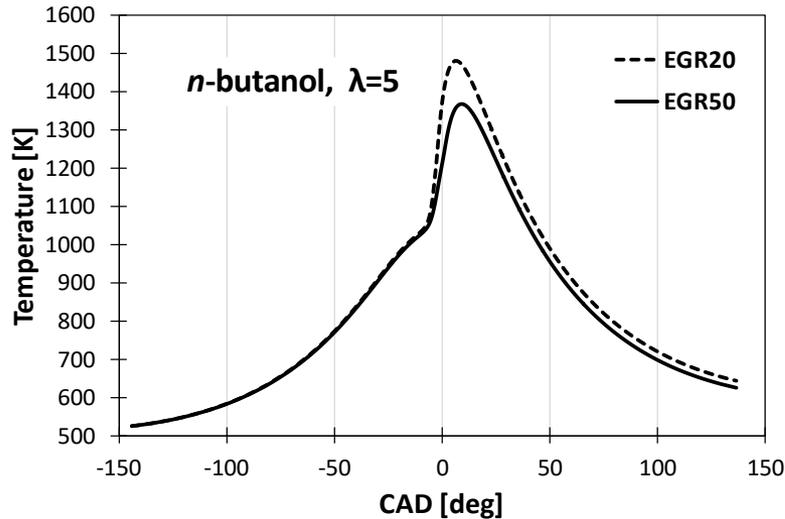
- Ethanol** and **TRF/butanol** extend the operability maps towards **ringing region** (highly anti-knocking components: toluene, butanol)
- n-butanol** and **n-pentanol** show the highest flexibility to load and EGR
- TRF/butanol** extends **partial burn limits** compared to ethanol and PRFs

TRF/butanol (RON=95, MON=87): *iso*-octane/*n*-heptane/toluene/*n*-butanol (41.2/8.1/22.4/28.3 mole%)

Agbro et al. *Fuel*, 2017

Multi-zone model of HCCI Engine: Operability Maps

Sensitivity Analysis to T, Inner Zone



Relative Branching of H-abstraction (OH, HO₂)

- Alkane moiety enhances ignition at LT
- Alcohol moiety inhibits ignition at LT

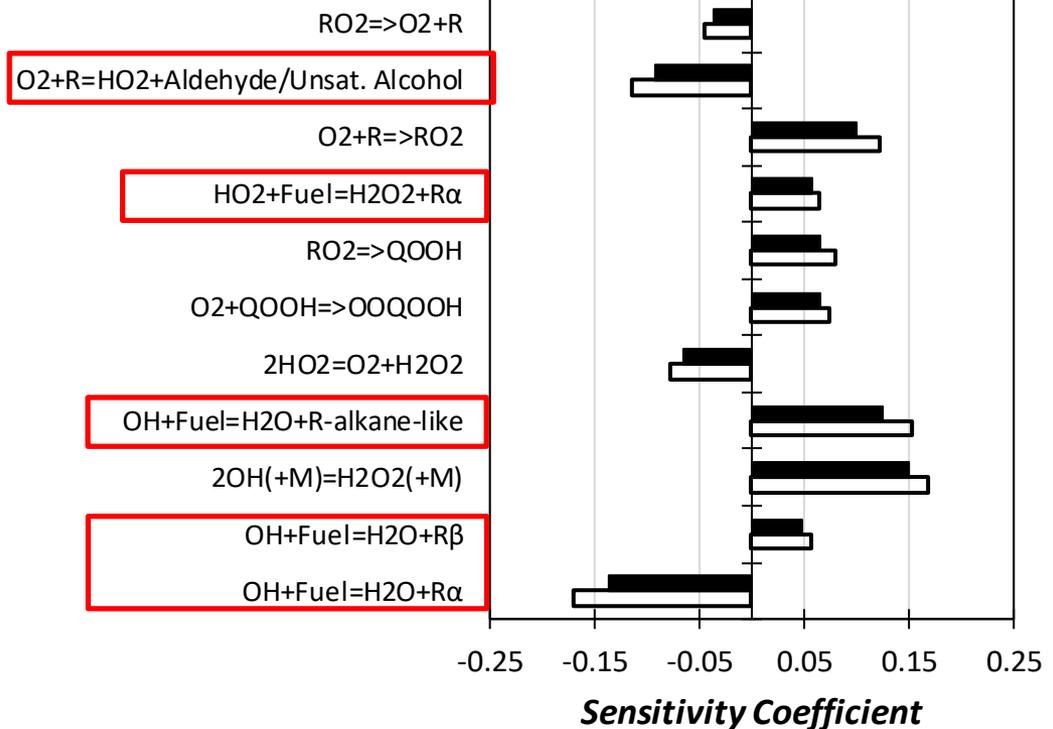
O₂+radicals=HO₂+aldehyde/enol

Only available for ethanol

- ✓ Da Silva et al. and Zador et al. provide basis: rate rule is needed!

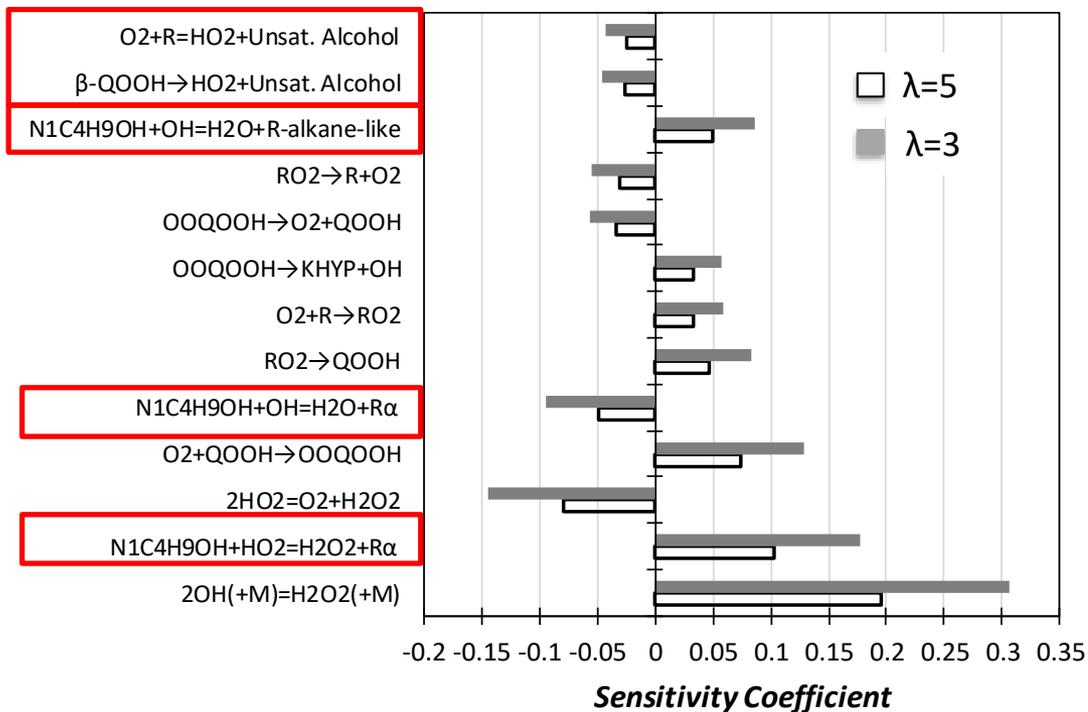
n-butanol, λ=5

■ EGR50 □ EGR20

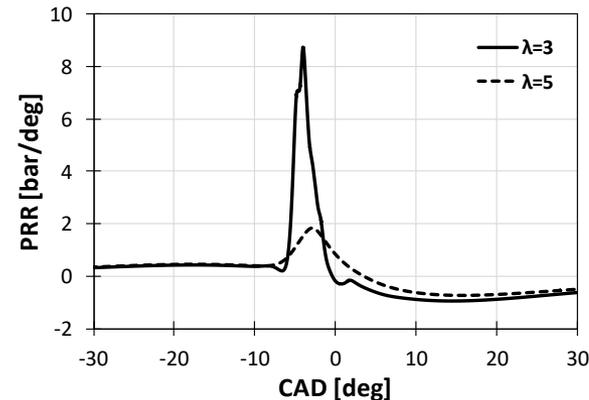
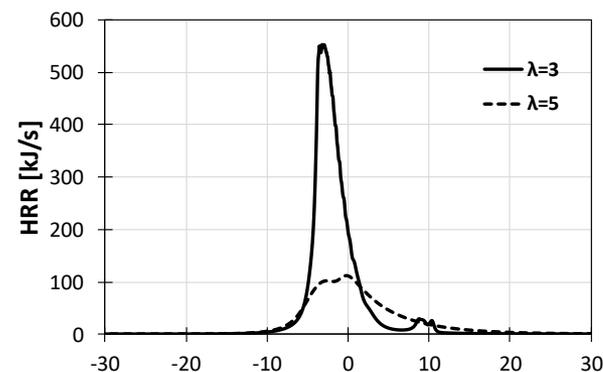
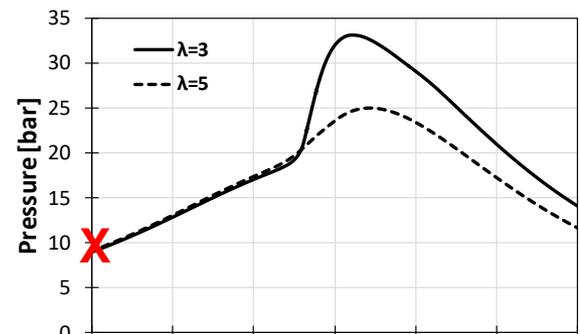


Multi-zone model of HCCI Engine: OpenSMOKE++ ICE Model

CAD ~ -30, p=9.1 atm T=904K

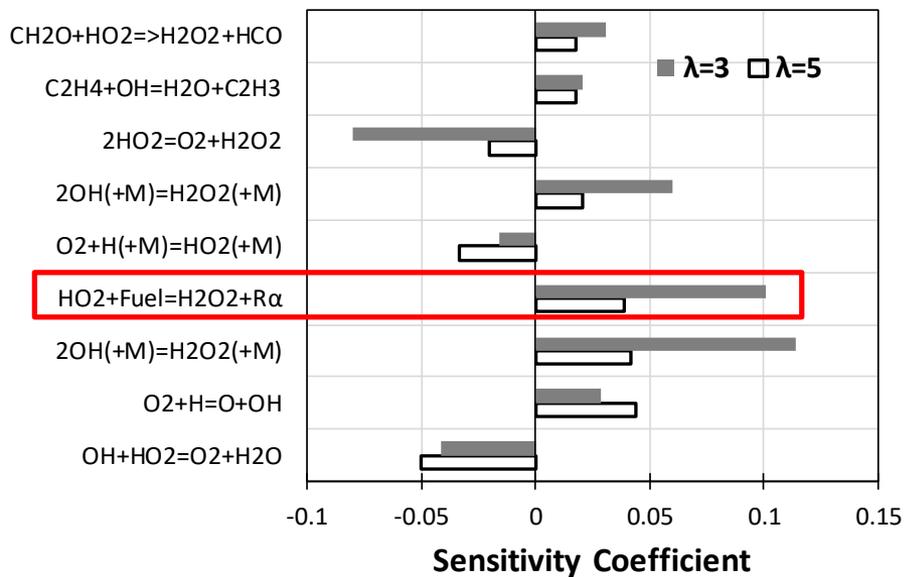


Mostly Fuel specific chemistry

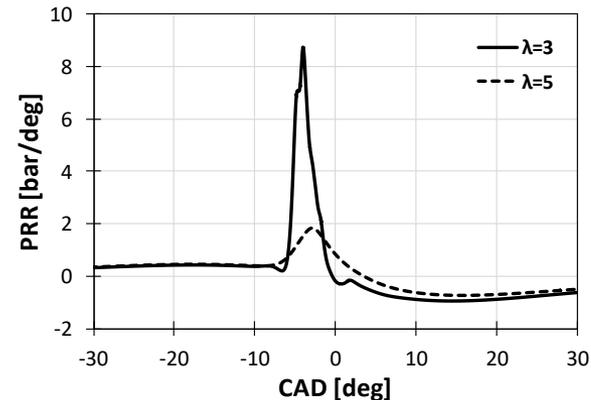
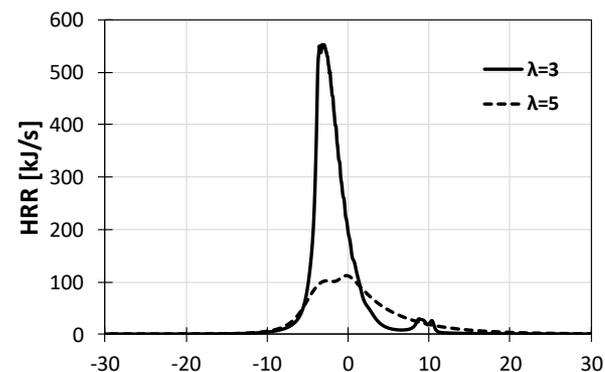
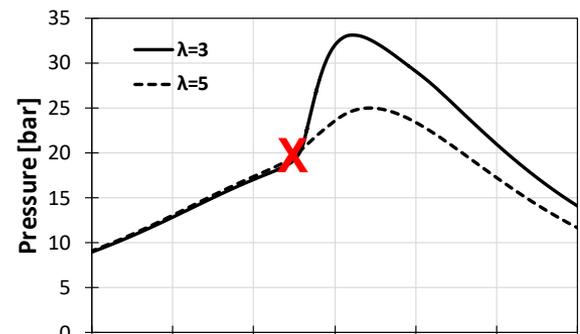


Multi-zone model of HCCI Engine: OpenSMOKE++ ICE Model

CAD ~ -5, p=18.9 atm, T=1034 K



Mostly C₀-C₂ and HO₂ chemistry



Conclusions

- **New Experimental Measurements in RCM and JSR** extend the scarce experimental data on alcohols oxidation at conditions relevant for real systems
- **POLIMI Kinetic Mechanism** has been **extended** to describe **the low temperature oxidation of alcohols (C₄-C₆) (with a new core mechanism!)**
- Coupling of the **POLIMI HCCI (ICE) Model** allows to **qualitatively explore the potential of alcohols as fuels or additives in HCCI Engine**, by means of **detailed kinetics**
- **Butanol and pentanol extend the operability region** allowing lower loads (i.e. reduce the partial burn region) compared to TRF/PRF mixtures and ethanol
- **Relevant kinetic pathways still deserve an accurate revision**, due to their strong impact on auto-ignition in real systems

Thank you!

Thank you!